Reaction Balancer

This is a calculator that balances chemical reactions. You're most welcome to download and use it, but please understand that it comes with no warranty whatsoever. Arithmetek Inc. accepts no liability whatsoever for its use, any errors, any consequences of using it, nor for any inability to use this calculator.

Having said that, I have found it a useful little tool at times, especially when I don't feel like figuring out stoichiometric coefficients myself, or when I need to calculate quantities of reagents, products, etc. from other input numbers. If you want it, just copy the zip file, save it somewhere convenient on your computer, extract the executable file and save it as something you'll recognize again. This is what it looks like when you first open it.

| ľ | Reaction Balanc | er | | | | | |
|---|---|---------------|---|-----------------|---|--|--|
| | Enter or modify the compounds involved here | | | | | | |
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| | | Click | <here balar<="" td="" to=""><td>nce the reactio</td><td>n</td></here> | nce the reactio | n | | |
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| | Compound | Molar Mass | Mole | Mass | | | |
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All you do is type the reactants and products you want into the upper box, separating them with blanks, commas or semi-colons. The elements do need to be put in correctly – it is case sensitive. (It'll work anyway sometimes, but if you put in FE or fe instead of Fe, for example, it'll give an error message.) It accepts brackets in formulas for individual compounds, but you do need to make sure that you've done them correctly. You also need to give it the correct compounds to balance – it doesn't check for consistency in the individual components. Here's an example of incorrectly entered input data. In this case, ferric sulphate is entered incorrectly. The next picture shows it done correctly.

| 📙 Reaction Balance | er | | | | | |
|--|---------------|---------------|-----------------|---|--|--|
| Enter or modify the compounds involved here | | | | | | |
| Fe SO4 Fe(SO)4)2 | | | | | | |
| | | | | | | |
| | Click | here to balar | ice the reactio | n | | |
| | | | | | | |
| The reaction matrix was solved, but the overall mass balance failed. Are all the compounds listed correct? | | | | | | |
| Compound | Molar Mass | Mole | Mass | | | |
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| 🖳 Reaction Balanc | er | | | | | | | |
|---|---------------|----------------|--------------------|---------------------------------------|--|--|--|--|
| Enter or modify the compounds involved here | | | | | | | | |
| Fe SO4 Fe2(SO4)3 | | | | | | | | |
| , | | | | | | | | |
| Click here to balance the reaction | | | | | | | | |
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| | | | | | | | | |
| | 2F | e+3SO4 = | Fe2(SO4)3 | | | | | |
| | (| | | | | | | |
| Compound | Molar Mass | Mole | Mass | Re-calculate | | | | |
| Reactants Fe | 55.847 | 2.000 | 444.604 | | | | | |
| SO4 | 96.064 | 2.000 3.000 | 111.694 288.191 | Changing any one value in the Mole | | | | |
| Products | | | | or Mass column | | | | |
| Fe2(SO4)3 | 399.885 | 1.000 | 399.885 | and clicking re- calculate bases | | | | |
| | | | | the values in both | | | | |
| | | | | columns on the number changed | | | | |
| | | | | | | | | |
| ' Reactant mass | - 300 885 | | , | | | | | |
| Product mass | | | | | | | | |
| | | | | | | | | |

In this example, I wanted to balance the reaction between iron and sulphate to get ferric sulphate. To make it ferrous sulphate instead, just change $Fe_2(SO4)_3$ to $FeSO_4$, as I've done in the next picture. You do need to know the chemistry you're calculating, as the calculator doesn't know whether it's been given Fe^{3+} or Fe^{2+} in this example – it only recognizes the element Fe.

| Enter or m | | | | | | | |
|---|---|--|--|--|--|--|--|
| Enter or modify the compounds involved here | | | | | | | |
| Fe SO4 FeSO4 | | | | | | | |
| | | | | | | | |
| Click here to balance the reaction | | | | | | | |
| 01101 | | | | | | | |
| | | | | | | | |
| | Fe + SO4 = | FeSO4 | | | | | |
| | | | | | | | |
| Molar | Mole | Mass | | | | | |
| Mass | | 11033 | Re-calculate | | | | |
| 55 847 | 1 000 | 55.847 | | | | | |
| 96.064 | 1.000 | 96.064 | Changing any one value in the Mole | | | | |
| | | | or Mass column | | | | |
| 151.911 | 1.000 | 151.911 | and clicking re- | | | | |
| | | | calculate bases the values in both | | | | |
| | | | columns on the | | | | |
| | | | number changed | | | | |
| | | | | | | | |
| 151.911 | | | | | | | |
| 151.911 | | | | | | | |
| | Molar Mass 55.847 96.064 151.911 | Fe + SO4 = Molar Mass Mole 55.847 1.000 151.911 1.000 151.911 | Mass Mole Mass 55.847 1.000 55.847 96.064 1.000 96.064 151.911 1.000 151.911 151.911 1.000 151.911 | | | | |

Once you've entered the various compounds properly and the calculator has balanced the reaction, there's a useful feature that allows you to change any number in the products or reagents and it'll recalculate the other numbers accordingly. In the next example I've used a reaction between water, sulphur dioxide and oxygen, giving sulphuric acid.

| 📙 Reaction Balance | er | | | | | | | |
|---|------------------------------------|----------------|------------------|---------------------------------------|--|--|--|--|
| Enter or modify the compounds involved here | | | | | | | | |
| H2O SO2 O2 H | H20 S02 02 H2S04 | | | | | | | |
| | | | | | | | | |
| | Click here to balance the reaction | | | | | | | |
| | | | | | | | | |
| | LIDO |) + SO2 + 0.50 | D2 - U2004 | | | | | |
| | HZC |) + SO2 + 0.50 | JZ = HZSU4 | | | | | |
| Commound | Molar | Mole | Mass | | | | | |
| Compound | Mass | wore | wass | Re-calculate | | | | |
| Reactants H2O | 18.015 | 1.000 | 18.015 | | | | | |
| SO2 O2 | 64.065 31.999 | 1.000 0.500 | 64.065 15.999 | Changing any one value in the Mole | | | | |
| | 51.555 | 0.000 | 10.000 | or Mass column and clicking re- | | | | |
| Products H2SO4 | 98.079 | 1.000 | 98.079 | calculate bases | | | | |
| | | | | the values in both columns on the | | | | |
| | | | | number changed | | | | |
| | | | | | | | | |
| Reactant mass | | | | | | | | |
| Product mass | . 98.079 | | | | | | | |
| | | | | | | | | |

You'll notice that, in the column headed Mole it lists the reactants as one molar unit of water, one of sulphur dioxide and half a molar unit of oxygen, giving one molar unit of sulphuric acid. In the column headed Mass it gives the same quantities, just as mass values. The molar values are all well and good, but if I wanted to make a ton of acid and had to buy the sulphur dioxide, water and oxygen, those being priced per kilogram, I'd want to have the masses of reagents in kilograms, per ton of acid. No problem, I simply change the number 98.079 in the Mass column to 1000 and click the re-calculate button.

| | Enter or m | odifv the con | npounds involve | ed here | | | |
|------------------------------------|---------------|---------------|-----------------|---------------------------------------|--|--|--|
| H2O SO2 O2 H: | 2504 | | | | | | |
| | | | | | | | |
| Click here to balance the reaction | | | | | | | |
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| | | | | | | | |
| | H20 |) + SO2 + 0.5 | 502 = H2SO4 | | | | |
| | | | | | | | |
| Compound | Molar Mass | Mole | Mass | Re-calculate | | | |
| Reactants H2O | 18.015 | 10 196 | 183.680 | | | | |
| SO2 | 64.065 | 10.196 | 653.193 | Changing any one value in the Mole | | | |
| 02 | 31.999 | 5.098 | 163.127 | or Mass column | | | |
| Products H2SO4 | 98.079 | 10 196 | 1000.000 | and clicking re- calculate bases | | | |
| H2504 | 90.079 | 10.190 | 1000.000 | the values in both | | | |
| | | | | columns on the number changed | | | |
| | | | | | | | |
| | | s out : 1000. | 000 | | | | |
| Mass in : 1000.0 | IIII Mag | | | | | | |

I have occasionally put in compounds that should be OK but the calculator thinks there's something wrong. The next picture shows example. (I used three blanks between each compound this time.)

In this example the calculator didn't read the component $H_3OFe_3(SO4)_2(OH)_6$ correctly. One day I'll get around to fixing that, but in all honesty it's not a high priority. If you're any good at C# and would like a go at it, please contact me and I'll give you the source code.

The picture after that shows an example in which there are not enough compounds for the equation to balance. I left out the sulphur dioxide.

The last picture shows an example in which I have too many compounds.

There you are. Keep your brain switched on when you use this calculator – it works quite well most of the time, but it is only a dumb algorithm...

Mike Dry.

| Reaction Balanc | er | | | | | |
|---|----------------|------------------|-----------------|---------------------|--|--|
| Enter or modify the compounds involved here | | | | | | |
| H2O Fe SO4 | H3OFe3(SO4 |)2(OH)6 H30 |) | | | |
| | | | | | | |
| | Click | here to balar | ice the reactio | n | | |
| | | | | | | |
| Element | : S appears or | ılv in SO4. It n | nust appear in | at least two of the | | |
| Lionion | . e appeare or | compounds | isted. | | | |
| | Molar | | | | | |
| Compound | Mass | Mole | Mass | | | |
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| 🖳 Reaction Balancer | | | | | | | | |
|---|--|------|------|--|--|--|--|--|
| Enter or modify the compounds involved here | | | | | | | | |
| H20 02 H2SC | 14 | | | | | | | |
| | | | | | | | | |
| | Click here to balance the reaction | | | | | | | |
| | | | | | | | | |
| Element | Element S appears only in H2SO4. It must appear in at least two of the compounds listed. | | | | | | | |
| Compound | Molar Mass | Mole | Mass | | | | | |
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| 📙 Reaction Balance | er | | | | | | |
|---|----------------------|------|------|--|--|--|--|
| Enter or modify the compounds involved here | | | | | | | |
| H20 02 S02 S | H2O O2 SO2 SO3 H2SO4 | | | | | | |
| | | | | | | | |
| Click here to balance the reaction | | | | | | | |
| | | | | | | | |
| The 5 compounds listed contain only 3 different atoms. You need at least 4 atoms to balance a reaction of 5 compounds. | | | | | | | |
| Compound | Molar Mass | Mole | Mass | | | | |
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